

Essential Physics II

List of equations covered in this course.

Equations in boxes will be provided in the exam. Please check you know where/when/how to use these equations.

Students are expected to **know** equations not in boxes.

Part I: Electromagnetism

$$\vec{F}_{12} = \frac{kq_1q_2}{r^2} \hat{r}_{12}$$

Coulomb's law

$$k = 9.0 \times 10^9 \text{ Nm}^2/\text{C}^2$$

$$\vec{E} = \frac{kq}{r^2} \hat{r}$$

$$\vec{E} = \sum_i d\vec{E} = \int \frac{k dq}{r^2} \hat{r}$$

$$p = qd$$

Dipole moment

$$\vec{\tau} = \vec{p} \times \vec{E}$$

Torque on dipole

$$U = -pE \cos \theta = -\vec{p} \cdot \vec{E}$$

Potential energy

$$\Phi = |\vec{E}| |\vec{A}| \cos \theta = \vec{E} \cdot \vec{A}$$

Electric flux

$$\oint \vec{E} \cdot d\vec{A} = \frac{q_{\text{enclosed}}}{\epsilon_0}$$

Gauss's Law

(know solutions for sphere, spherical shell, line, plane and conductors)

$$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{Nm}$$

$$\Delta V_{AB} = \frac{\Delta U_{AB}}{q}$$

Electric potential difference

$$\Delta V_{AB} = - \int_A^B \vec{E} \cdot d\vec{r} = -\vec{E} \cdot \Delta\vec{r}$$

$$V(r) = \frac{kq}{r}$$

Point charge potential

$$\vec{F}_B = q\vec{v} \times \vec{B}$$

Magnetic force

Unit: [T] Tesla

$$\vec{F}_B = q\vec{E} + q\vec{v} \times \vec{B}$$

Electromagnetic force

$$\vec{F} = I\vec{l} \times \vec{B}$$

Magnetic force on a current

$$d\vec{B} = \frac{\mu_0}{4\pi} \frac{Id\vec{l} \times \vec{r}}{r^2}$$

Biot-Savart Law
(know solutions for current loop, straight wire)

$$\mu_0 = 4\pi \times 10^{-7} \text{ N/A}^2$$

$$\vec{B} = \int d\vec{B} = \frac{\mu_0}{4\pi} \int \frac{Id\vec{l} \times \hat{r}}{r^2}$$

$$\vec{\mu} = NI\vec{A}$$

Magnetic dipole moment

$$\vec{\tau} = \vec{\mu} \times \vec{B}$$

Torque on a current loop

$$\oint \vec{B} \cdot d\vec{A} = 0$$

Gauss's law for magnetism

$$\oint \vec{B} \cdot d\vec{r} = \mu_0 I_{\text{enclosed}}$$

Ampere's law for steady current (know solutions for inside and outside a wire, current sheet and solenoid)

$$\mathcal{E} = IR$$

Ohm's law

\mathcal{E} electromotive force, EMF

$$\Phi_B = \int \vec{V} \cdot d\vec{A} = \vec{B} \cdot \vec{A}$$

Magnetic flux

$$\oint \vec{E} \cdot d\vec{r} = -\frac{d\Phi_B}{dt}$$

Faraday's Law
(know solutions for a solenoid)

$$\mathcal{E} = -\frac{d\Phi_B}{dt}$$

$$\oint \vec{B} \cdot d\vec{r} = \mu_0 I + \mu_0 \epsilon_0 \frac{d\Phi_E}{dt}$$

Ampere's law with Maxwell modification

$$S = S_0 \cos^2 \theta$$

Law of Malus (Polarisation)

Part II: Thermodynamics

$$T_C = T - 273.15$$
 Celsius to kelvin scale

$$T_F = \frac{9}{5}T_C + 32$$
 Fahrenheit to celsius scale

$$\Delta Q = mc\Delta T$$
 Heat capacity c specific heat capacity

$$H = \frac{dQ}{dt} = -kA \frac{dT}{dx}$$
 Conductive heat flow

$$H = \frac{T_1 - T_3}{R_1 + R_2}$$
 Composite slab

$$R = \frac{\Delta x}{kA}$$
 Thermal resistance

$$P = \epsilon\sigma AT^4$$
 Stefan-Boltzmann Law $\sigma = 5.67 \times 10^{-8} \text{ Js}^{-1}\text{m}^{-2}\text{K}^{-4}$

$$pV = nRT$$
 Ideal gas law $R = 8.314 \text{ J/K} \cdot \text{mol}$
 $N_A = 6.023 \times 10^{23} \text{ atoms}$

$$\frac{3}{2}kT = \frac{1}{2}m\bar{v}^2$$
 Kinetic theory of gases

$$v_{\text{th}} = \sqrt{\frac{3kT}{m}}$$
 Thermal speed
(typical molecular speed)

$$Q = Lm$$
 Energy required for a phase change L heat of transformation

$$\Delta U = Q + W$$
 1st Law of Thermodynamics

$$\frac{dU}{dt} = \frac{dQ}{dt} + \frac{dW}{dt}$$

$$Q = -W = nRT \ln \left(\frac{V_2}{V_1} \right)$$
 Isothermal process
(T = constant)

$$\Delta U = nC_v\Delta T$$
 Any process

$$W = -p\Delta V \quad \text{Isobaric gas (p = constant)}$$

$$Q = nC_P\Delta T = nC_V\Delta T + p\Delta V$$

$$C_P = C_V + R$$

molar specific heat

$$\Delta U = W$$

Adiabatic process (Q = 0)

$$pV^\gamma = \text{constant}$$

$$\gamma = \frac{C_P}{C_V}$$

$$TV^{\gamma-1} = \text{constant}$$

$$W = \frac{p_2V_2 - p_1V_1}{\gamma - 1}$$

$$\frac{1}{2}kT$$

Average energy / molecule for each degree of freedom

Part III: Modern Physics

$\lambda_{\text{peak}}T = 2.898 \text{ mm} \cdot \text{K}$	Wien's law	
$P_{\text{blackbody}} = \sigma AT^4$	Stefan-Boltzmann law	$\sigma = 5.67 \times 10^{-8} \text{ Js}^{-1}\text{m}^{-2}\text{K}^{-4}$
$R(\lambda, T) = \frac{2\pi ckT}{\lambda^4}$	Rayleigh-Jean law	
$R(\lambda, T) = \frac{2\pi hc^2}{\lambda^5 (e^{hc/\lambda kT} - 1)}$	Planck's law	
$E = hf$		
$K_{\text{max}} = hf - \phi$	Photoelectric effect	
$\frac{1}{\lambda} = R_H \left(\frac{1}{n_2^2} - \frac{1}{n_1^2} \right)$	Hydrogen spectrum	$R_H = 1.097 \times 10^7 \text{ m}^{-1}$
$r = n^2 a_0$	Allowed radii in Bohr atom	$a_0 = 0.0529 \text{ nm}$
$E = -\frac{ke^2}{2a_0} \left(\frac{1}{n^2} \right)$	Energy levels in Bohr atoms	
$\lambda = \frac{h}{p}$	de Broglie wavelength	
$\Delta x \Delta p \geq \hbar$	Uncertainty principal	
$-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + U(x)\psi(x) = E\psi(x)$	Schrodinger equation	$\hbar = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$
$P(x) = \psi^2(x)dx$	Probability of detecting a particle	
$E = \frac{n^2 h^2}{8mL^2}$	Energy levels in an infinite square well	
$E = \left(n + \frac{1}{2} \right) \hbar\omega$	Energy levels in a harmonic oscillator	

